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about 1 kg/m^3 so we
get $m = \rho V = 1 \text{ kg/m}^3$
 $\times 0.001 \text{ m}^3 = 0.001$

kg
A more accurate
value from Table A.5
is $\rho = 1.17 \text{ kg/m}^3$ at
100 kPa, 25°C .

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 $= \rho V = 110 \text{ kg/m}^3 \times$
 $100\,000 \text{ m}^3 = 11 \times$

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10 6 kg Just to put
this in perspective a.

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The equation for a
liquid is: $v = \text{Constant}$
 $= v_0$. If you include
that v increases a little
with T then: $v = v_0 + C$
 $(T - T_0)$ where C is a
small constant with

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has a cross sectional
area of 15 ft². At the
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Solution: $P = \rho gh$; P_0
 $= (1.025/1.01325)$

$\times 14.696 = 14.866$

lbf/in^2 $P_{\text{ocean}} = P_0 +$

$P = 14.866 + P_{\text{hill}} =$

$P_0 - P = 14.866 -$

$62.3 \times 30 \times g = 27.84$

lbf/in^2 $g_c \times 144$

$0.0735 \times 300 \times g =$

14.71 lbf/in^2 $g_c \times 144$

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Solution: $m = \rho V =$
 $110 \text{ kg/m}^3 \times 100\,000$
 $\text{m}^3 = 11 \times 10^6 \text{ kg}$

Just to put this in perspective a power plant that generates 2000 MW by burning coal would make about 20 million tons of carbon-dioxide a

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Solution: $\rho_{\text{Hg}} : L = 1$
 $m; \rho = 13\,580 \text{ kg/m}^3$

from Table A.4 (or
read Fig 2.7) The
pressure difference
 ΔP balances the
column of height L so
from Eq. 2.2 $\Delta P = \rho g L$
 $= 13\,580 \text{ kg/m}^3 \times$
 $9.80665 \text{ m/s}^2 \times 1.0 \text{ m}$
 $\times 10^{-3} \text{ kPa/Pa} = 133.2$
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Wylen 2.14 You dive 5
m down in the ocean.

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